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Chemistry 702: Percentage Composition and Empirical ...

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Problem set #1: (on back) - Georgia Public Broadcasting

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CHEMISTRY: A Study of Matter. © 2004, GPB. 7.15. Example

Problem: Find the % by mass of oxygen in water. Percentage by mass of element in a compound = (mass of element in 1 mol of compound ÷ molar mass of compound) x 100 %. (after completing lab #1) Find the % of carbon in sodium bicarbonate (NaHCO₃).

Example Problem: Find the % by mass of oxygen in water.

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Physics 702: Torque and Rotational Equilibrium | Georgia

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Percentage by mass

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Lab One: Episode 702—Datasheet Name

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of Matter © 2004, GPB 7.1 Lab results: 1 doz grains of rice = _____g (Use this fact as a conversion

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mass of empty test tube mass of test tube with copper ...

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