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7-2 Review and Reinforcement. Covalent Bonding. If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. ____F, covalent_ 1. A group of atoms united by ionic bonds is called a molecule. ____T____ 2. A covalent bond is formed by a shared pair of ...

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7-2 Review and Reinforcement (continued) Write Lewis structures for each of the following molecules. Indicated the bonds with either dots or dashes. 15. NH₃ 18. PCl₃. 16. H₂ 19. CCl₄. 17. C₂H₄ 20. C₂H₆. Answer the following questions as directed. 21. Explain why the molecules SF₄ is an exception to the octet rule.\

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6—2 Review and Reinforcement (continued) Identify each of the following elements as a meta/ (M), nonmetal (NM), or semimetal (SM). 19. sodium —SM— silicon 21. neon 22. 23. nitrogen Write the family names that have been given to each of the following groups. 24. Group IA alkan- 25. Group 2A 26. Group 7 A 27. Group8A hvohlt

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